

Calculating subsurface nucleonic production of noble gas nuclides: implications on crustal and mantle K, Th, U abundances

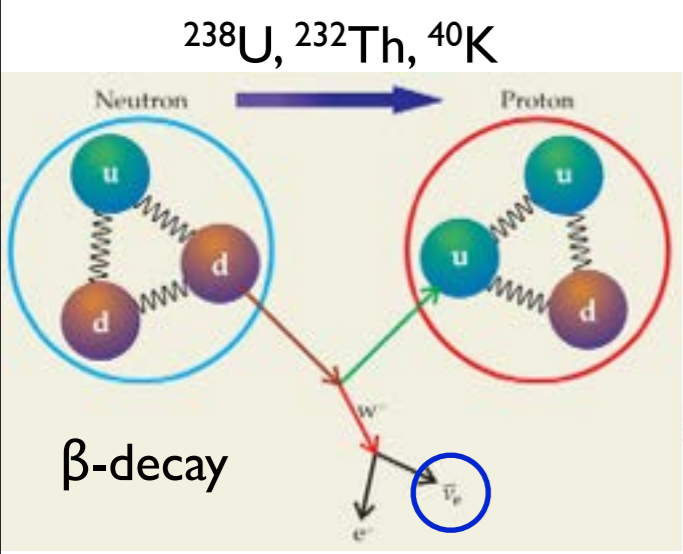


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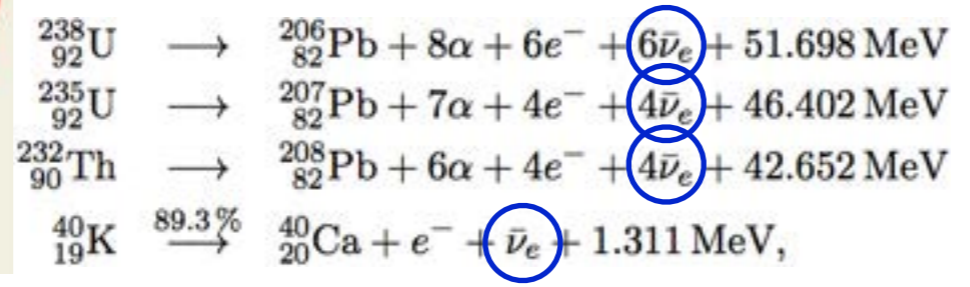


Collaboration with
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Sujoy Mukhopadhyay (Harvard)
Lauren Stevens (Univ. Maryland)
Jerry Peterson (CU Boulder)

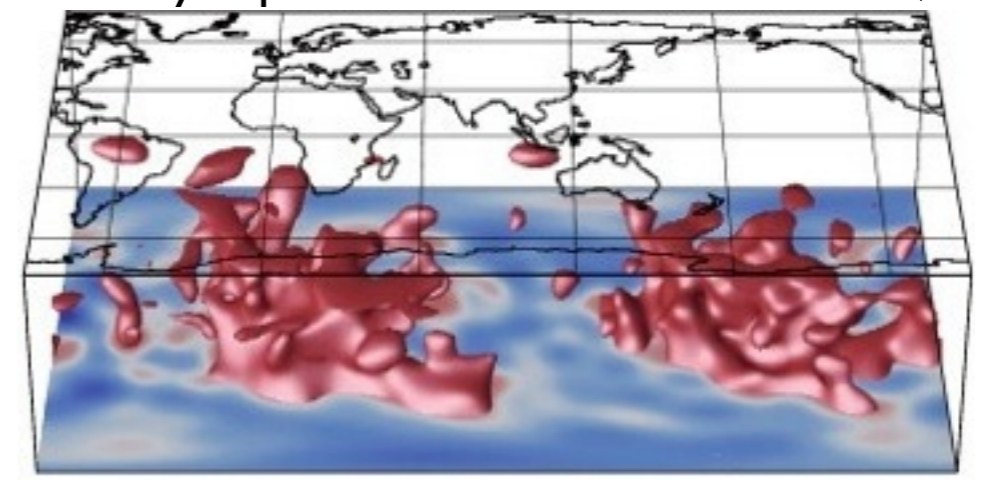
How much radiogenic heating in the mantle??
 What is the Earth made of??
 Chemical reservoirs in the mantle??



Geoneutrinos:
 electron antineutrinos emitted in β -decays of natural radionuclides

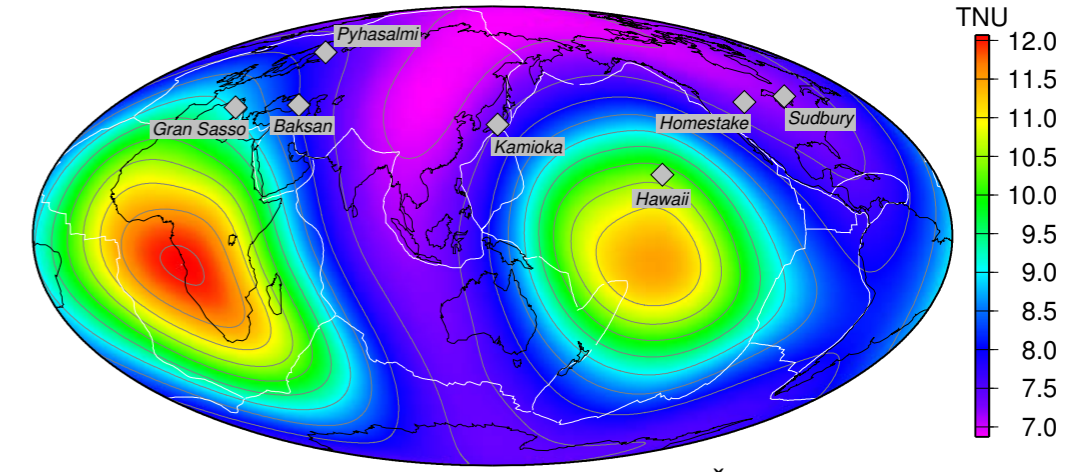


LLSVPs may represent material enriched in U, Th, K



from Bull et al. 2009

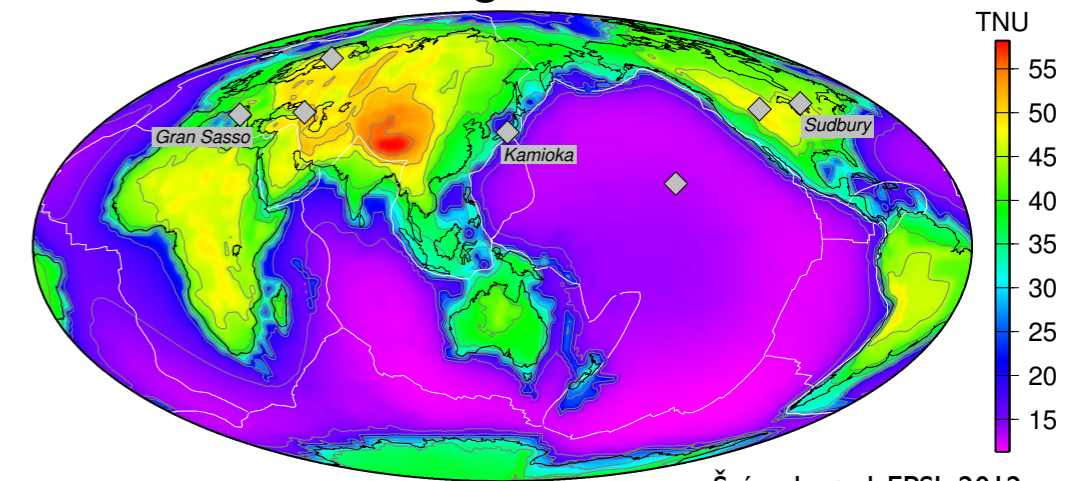
Predicted mantle geoneutrino flux



Šrámek et al. EPSL 2012

Mantle: detectable variation!
Must measure in the ocean...

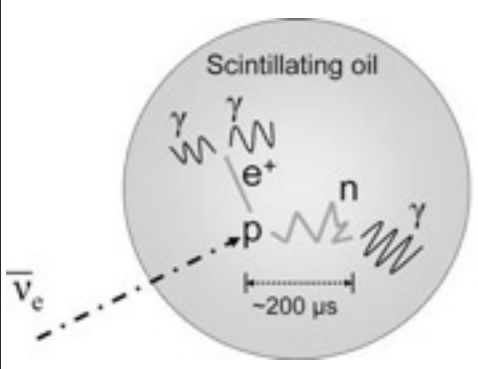
Crust + mantle geoneutrino flux



Šrámek et al. EPSL 2012

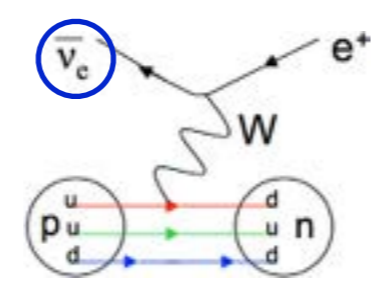
Geo-ν detection possible:

Large liquid scintillator detectors

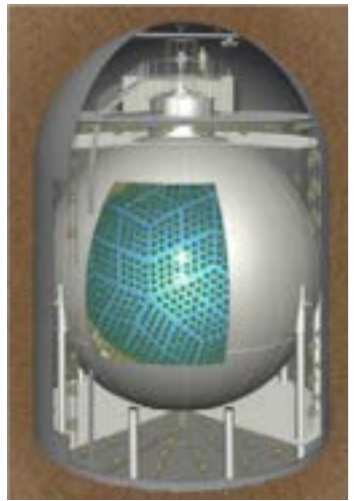


Borexino (Italy)

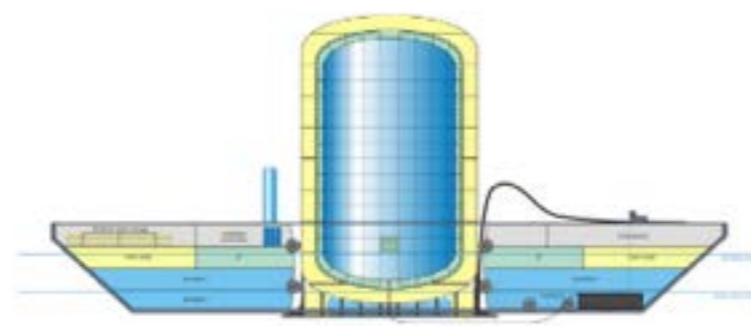
Direct information on Earth's deep-seated radioactivity!



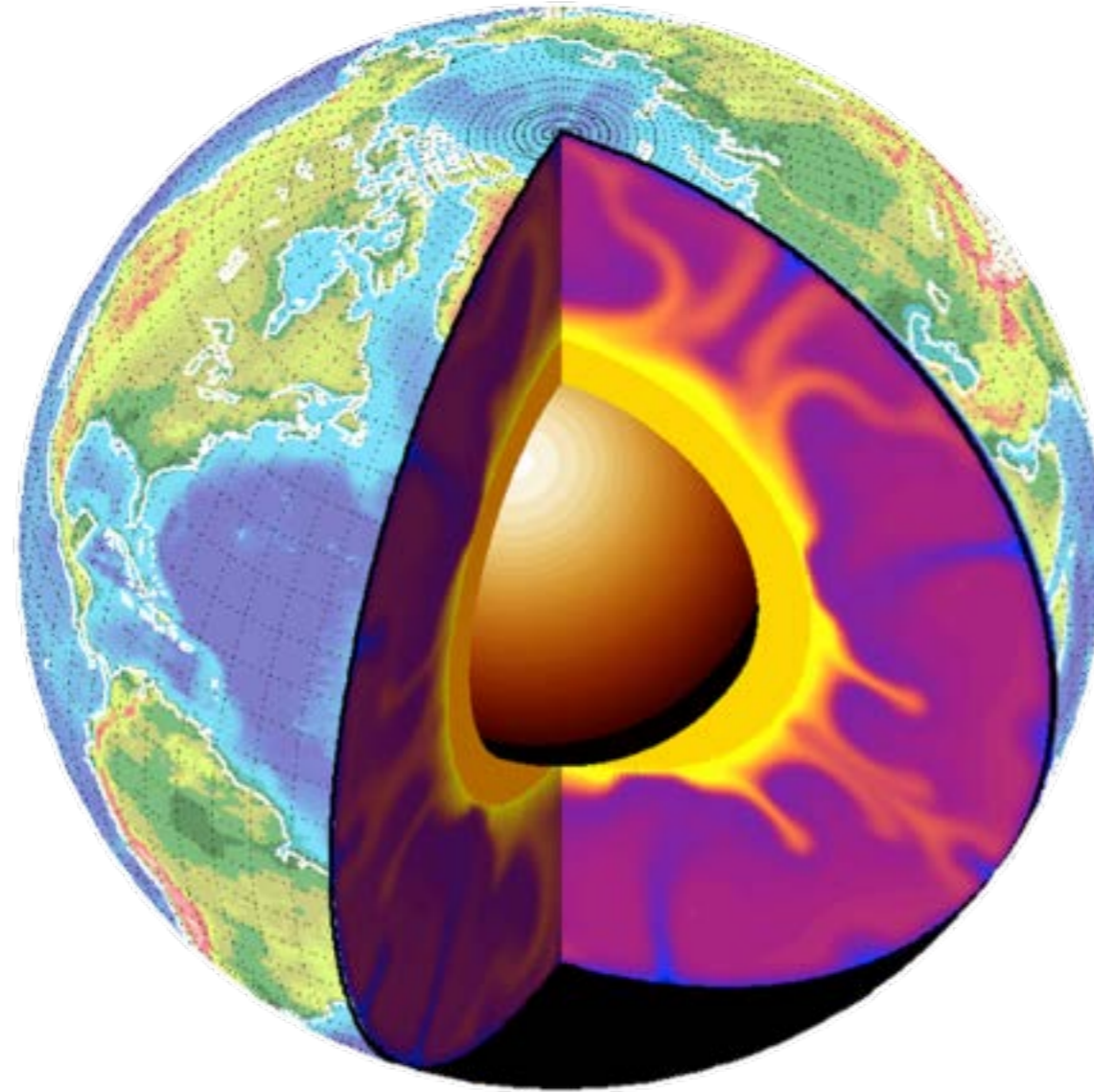
KamLAND (Japan)



Hanohano, proposed ocean deployment

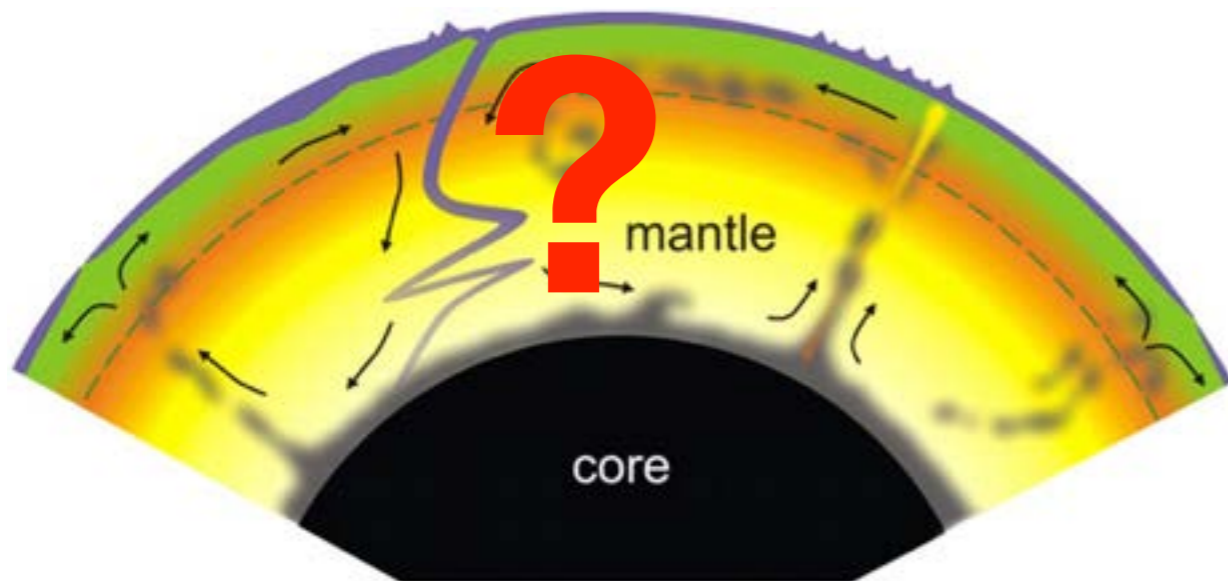


How much U, Th, K in the Earth?



Estimates range from 9 to 36 TW radiogenic power

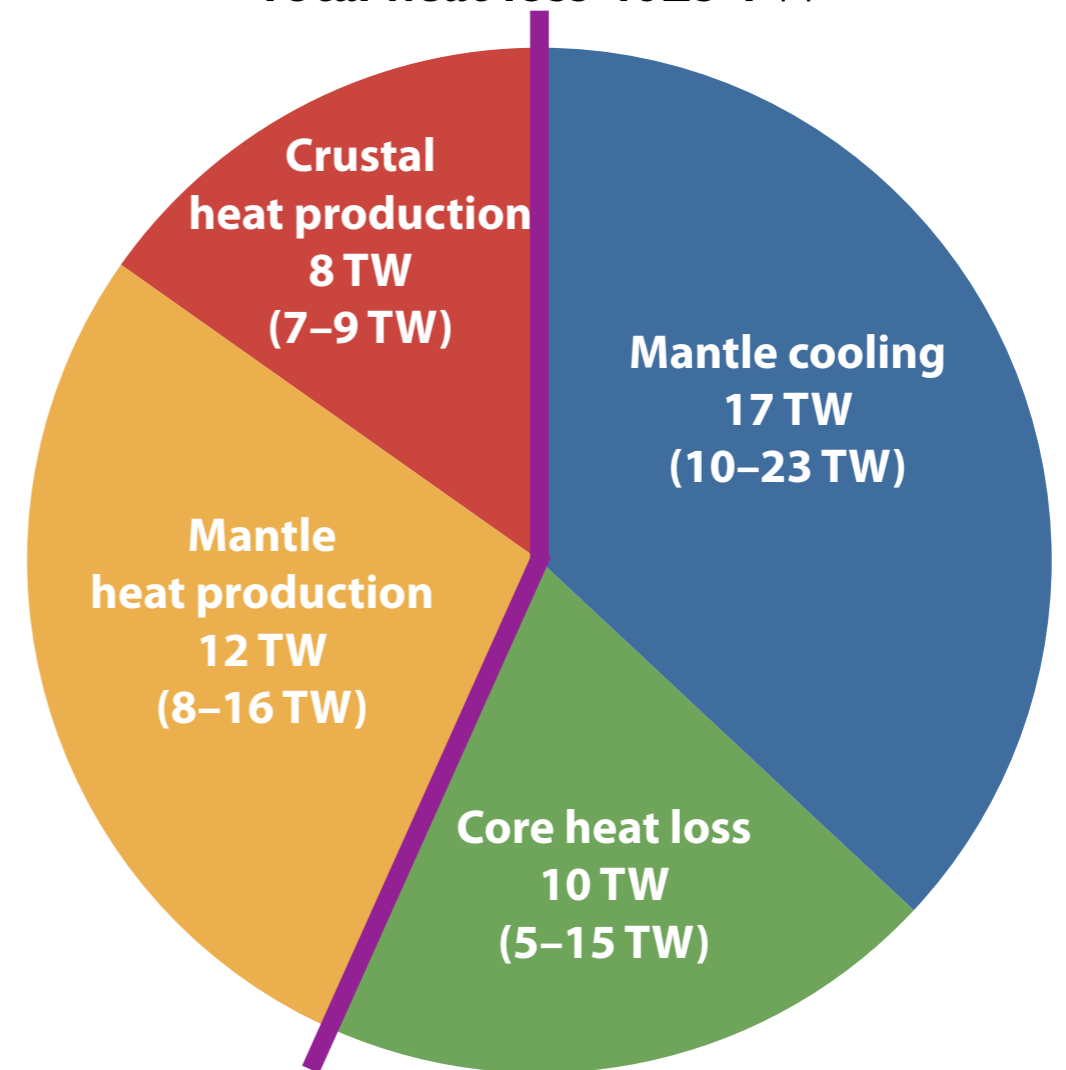
How much radiogenic heating in the mantle to power convection?



Estimates from 2 to 29 TW radiogenic power in the mantle

Earth's energy budget

Total heat loss 46 ± 3 TW



radiogenic + primordial heat + other...

U Th K

Composition of Silicate Earth (BSE)

TW radiogenic power
BSE **Mantle**

- **“Geochemical” estimate**
 - Ratios of RLE abundances constrained by C1 chondrites
 - Absolute abundances inferred from Earth rock samples
 - *McDonough & Sun (1995), Allègre (1995), Hart & Zindler (1986), Palme & O’Neill (2003), Arevalo et al. (2009)*
- **“Cosmochemical” estimate**
 - Isotopic similarity between Earth rocks and E-chondrites
 - Build the Earth from E-chondrite material
 - *Javoy et al. (2010)*
 - also “collisional erosion” models (*O’Neill & Palme 2008*)
- **“Geodynamical” estimate**
 - Based on a classical parameterized convection model
 - Requires a high mantle Urey ratio, i.e., high U, Th, K

20±4

13±4

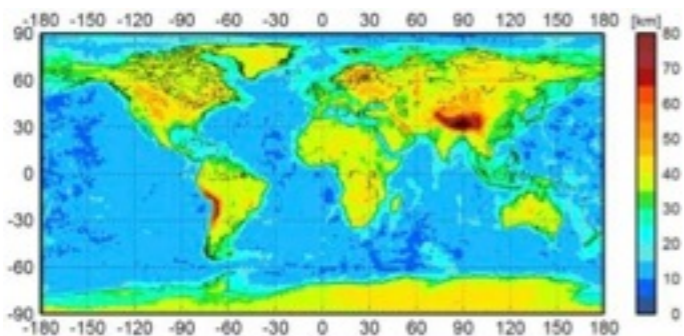
11±2

4±2

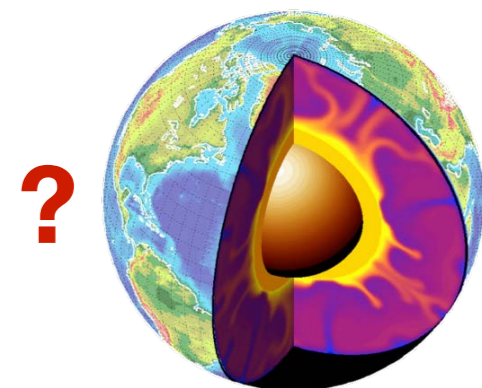
33±3

26±3

BSE = Mantle + Crust



Oceanic: 0.22 ± 0.03 TW
 Continental: 6.8 (+1.4/−1.1) TW



Approach:

Use noble gas isotopic ratios
from gases originating at depth

- Why now?
- How to calculate ^{39}Ar production rate
- Some results

Isotopes of Argon

34 known isotopes

^{40}Ar ... radiogenic, stable

^{39}Ar ... cosmo/nucleogenic, $t_{1/2} = 269 \text{ y}$

^{36}Ar ... primordial, stable

Atmosphere

^{40}Ar from degassing of Earth over 4.5 Gy

^{39}Ar produced cosmogenically from ^{40}Ar

$^{40}\text{Ar}/^{36}\text{Ar} = 295$

$^{39}\text{Ar}/^{40}\text{Ar} = 8 \times 10^{-16}$

Underground

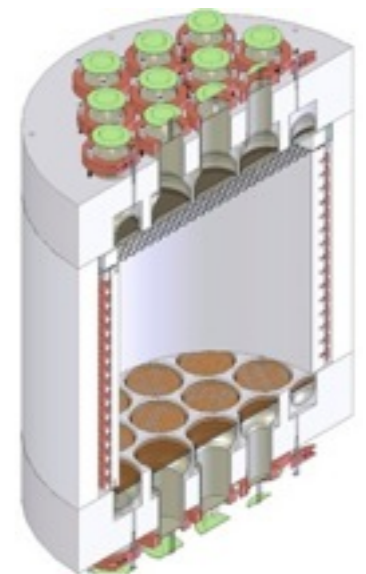
^{40}Ar produced by electron capture on ^{40}K

^{39}Ar produced nucleogenically from ^{39}K

$^{39}\text{Ar}/^{40}\text{Ar} < 0.006$ atmospheric

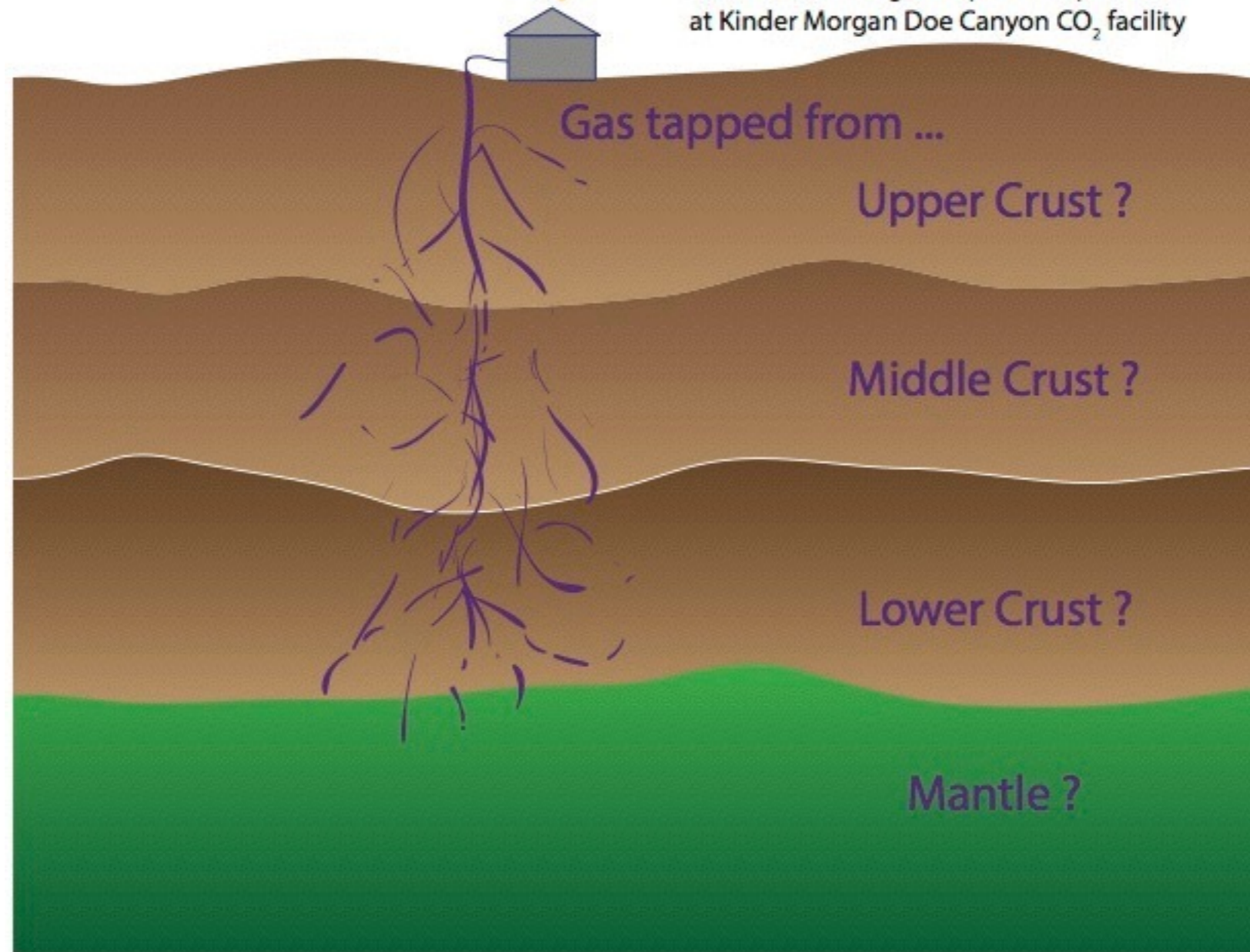
Dark matter WIMP search

- Dark matter detectors looking for Weakly Interacting Massive Particles (WIMPs) require low radioactivity argon
- Atmospheric level ($^{39}\text{Ar}/^{40}\text{Ar}=8\times 10^{-16}$) is too high
- Gas from deep CO_2 wells shows lower level of ^{39}Ar (e.g., Cortez CO, Bueyeros NM)
- $^{39}\text{Ar}/^{40}\text{Ar}$ – challenging measurements...
 - low-level radioactive decay counting
 - Atomic Trap Trace Analysis (ATTA)

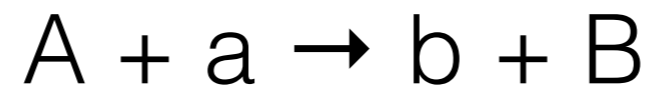




DarkSide's argon separation plant
at Kinder Morgan Doe Canyon CO₂ facility



Nuclear physics notation:



A ... target nuclide

a ... projectile

b ... ejectile

B ... product nuclide

(α ,n)

(n,p)

(n, α)

(n,2n)

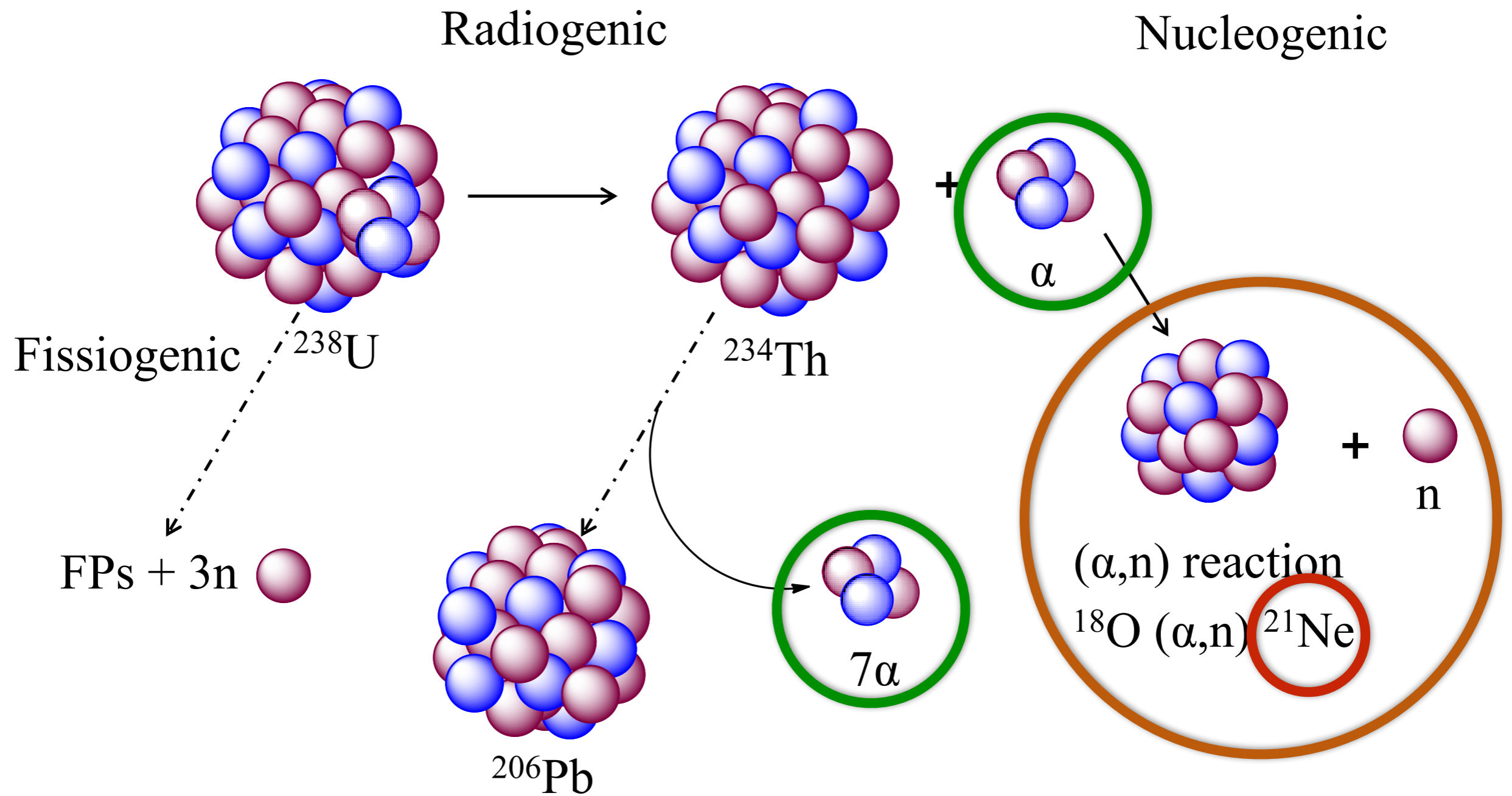
...

$^{40}\text{Ar}(n,2n)^{39}\text{Ar}$
atmosphere, cosmogenic

Nucleogenic production

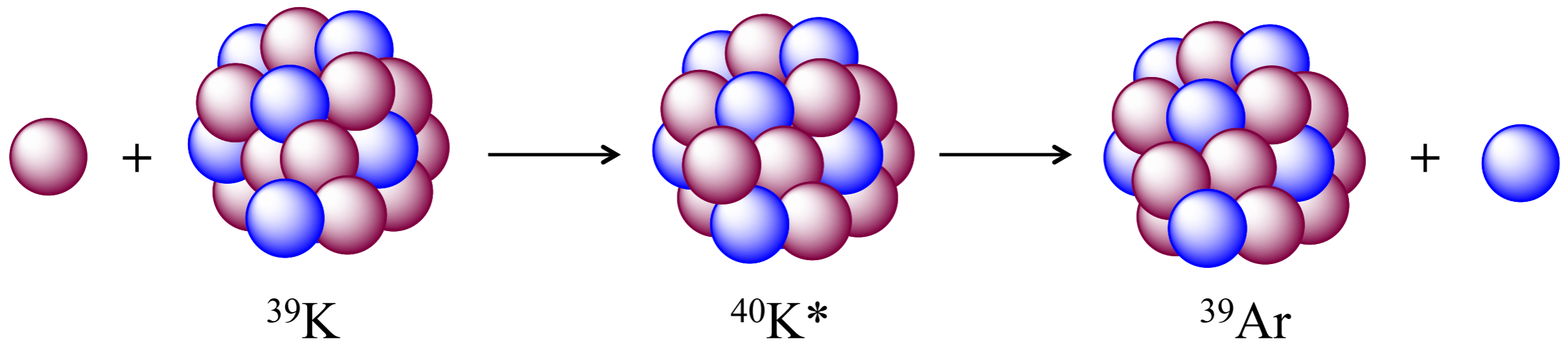
- Decay of radioactive U and Th in Earth's interior produce α particles
- (α , n) reactions on light isotopes produce neutrons
- Neutrons are also produced by spontaneous fission of U
- (n,p) reaction then produces ^{39}Ar from ^{39}K
- \Rightarrow Measurement of isotopic ratios in outgassing rock can inform us about the U, Th, K

1. U, Th decay produces α 's



2. (α, n) produce neutrons

$^{39}\text{K}(n,p)$ produces ^{39}Ar



composite
nucleus

Noble gas isotopic ratios

- ^{21}Ne production rate proportional to $[\text{U}+\text{Th}]$
- ^{39}Ar production rate proportional to $[\text{K}] \times [\text{U}+\text{Th}]$
- ^{40}Ar production rate proportional to $[\text{K}]$

therefore

- $^{39}\text{Ar}/^{40}\text{Ar}$ proportional to $[\text{U}+\text{Th}]$
- $^{39}\text{Ar}/^{21}\text{Ne}$ proportional to $[\text{K}]$
- $^{40}\text{Ar}/^{21}\text{Ne}$ proportional to $[\text{K}]/[\text{U}+\text{Th}]$

Calculating ^{39}Ar nucleogenic production rate

- α emission from natural radionuclides
- neutron production by α -induced reactions
- $^{39}\text{K}(n,p)^{39}\text{Ar}$

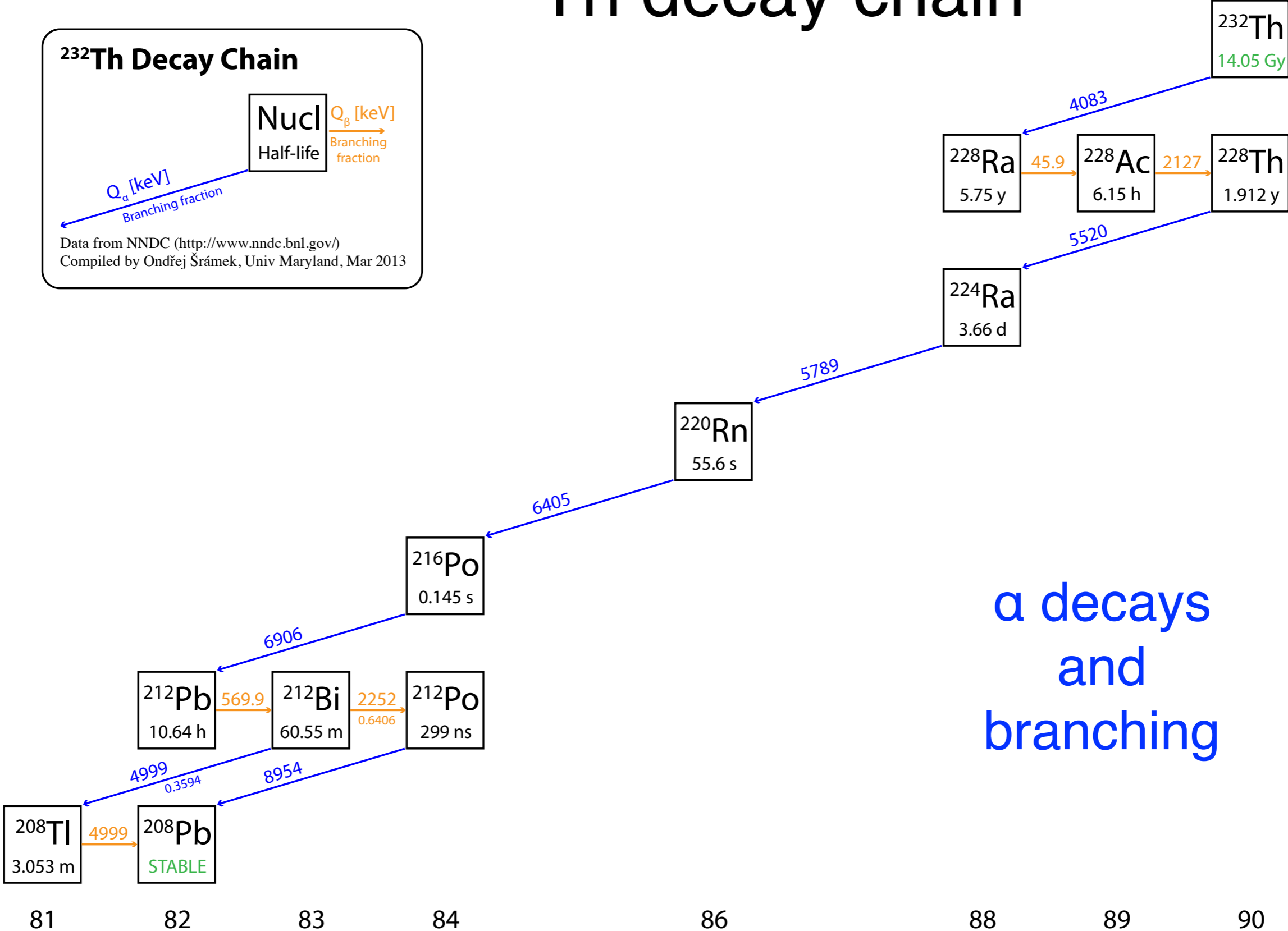
Table 7: ^{39}Ar production rates as calculated in several studies. Rates are recalculated to a common K, Th, U composition of Upper Crust in *Rudnick and Gao (2003)*, K=2.3 %, Th=10.5 ppm, U=2.7 ppm by weight. ^{39}Ar prod. rate in number of atoms per year per kg of rock.

Reference	^{39}Ar prod. rate
<i>Mei et al. (2010)</i>	11
<i>Yokochi et al. (2012)</i>	55
<i>Yokochi et al. (2013)</i>	170; 110
This study	30

^{232}Th decay chain

^{232}Th Decay Chain

Data from NNDC (<http://www.nndc.bnl.gov/>)
 Compiled by Ondřej Šrámek, Univ Maryland, Mar 2013



α decays
and
branching

^{238}U decay chain

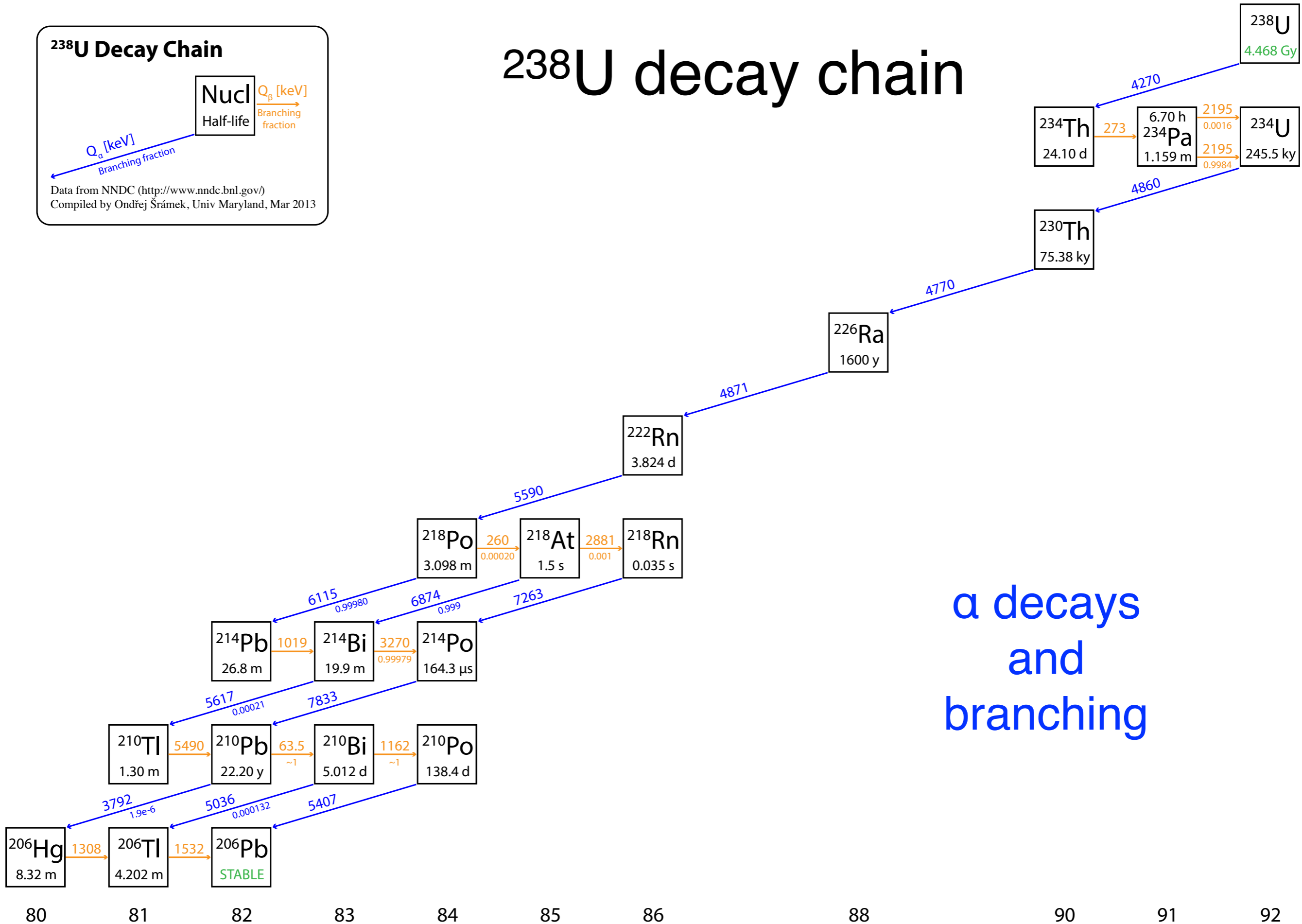
^{238}U Decay Chain

Nucl
 Half-life

Q_β [keV]
 Branching fraction

Q_α [keV]
 Branching fraction

Data from NNDC (<http://www.nndc.bnl.gov/>)
 Compiled by Ondřej Šrámek, Univ Maryland, Mar 2013



α decays
and
branching

235U decay chain

235U Decay Chain

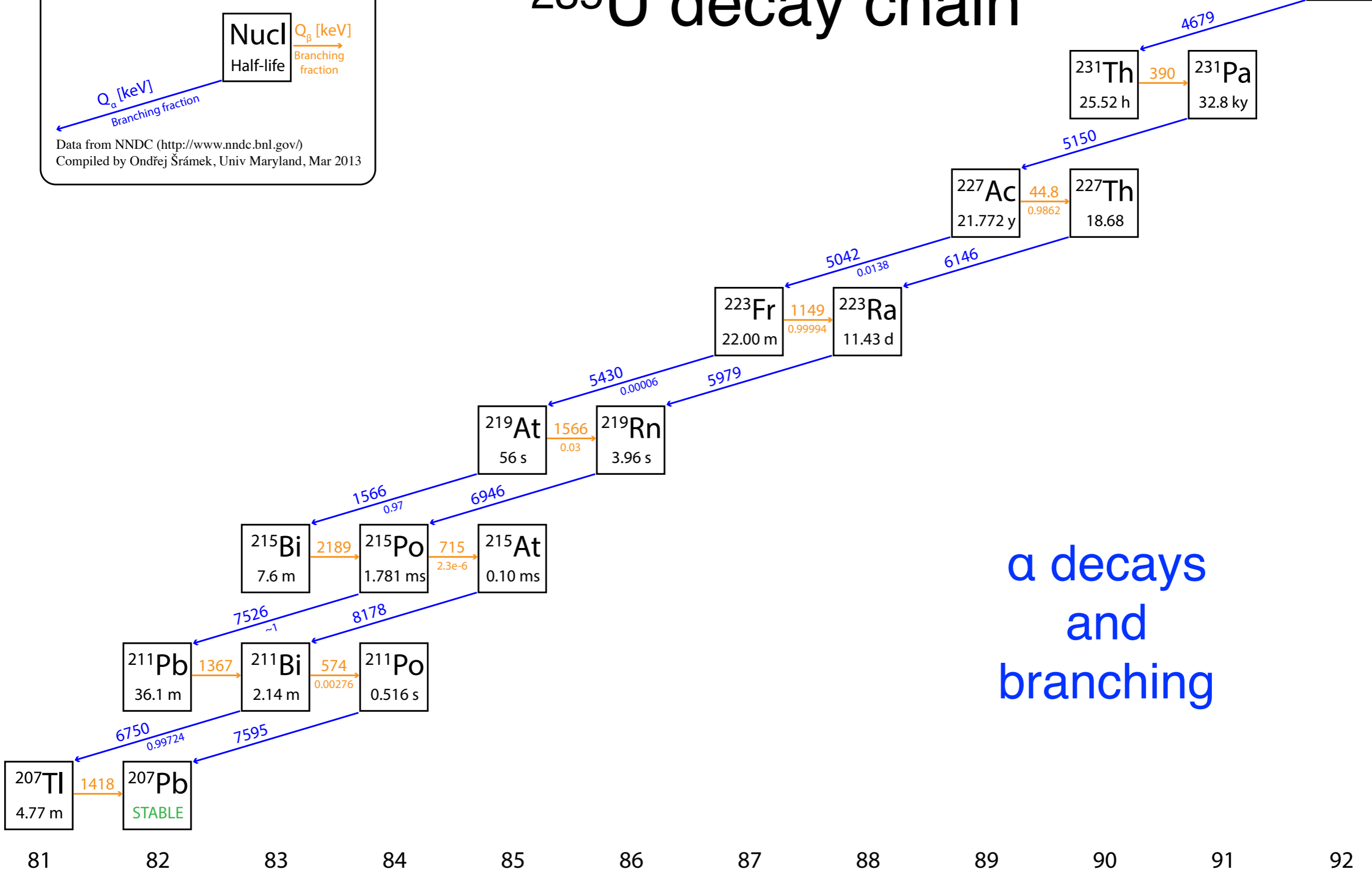
Q_α [keV]
 Branching fraction

Q_β [keV]
 Branching fraction

Nucl
 Half-life

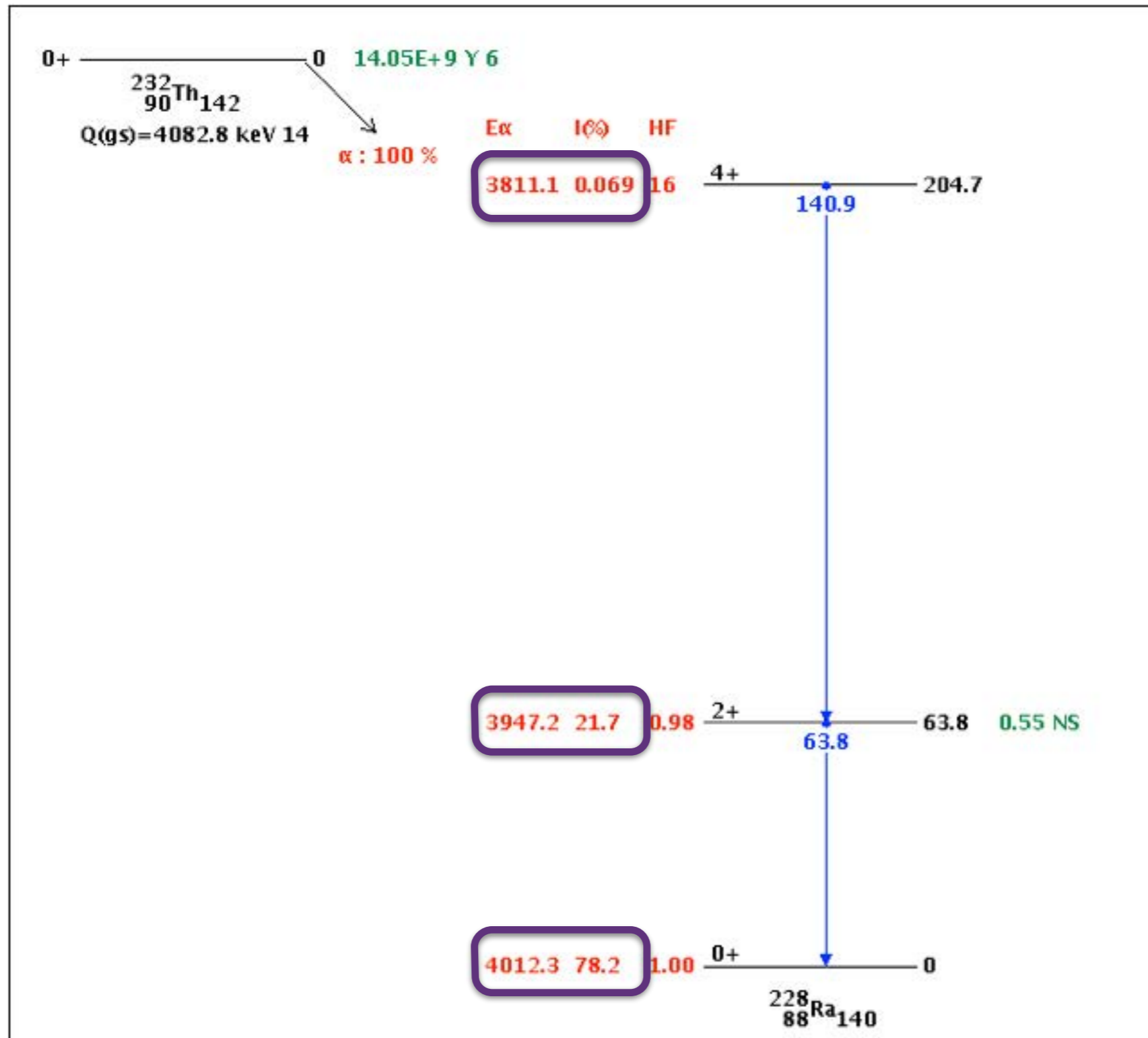
Data from NNDC (<http://www.nndc.bnl.gov/>)
 Compiled by Ondřej Šrámek, Univ Maryland, Mar 2013

235U
703.8 My



α decays
and
branching

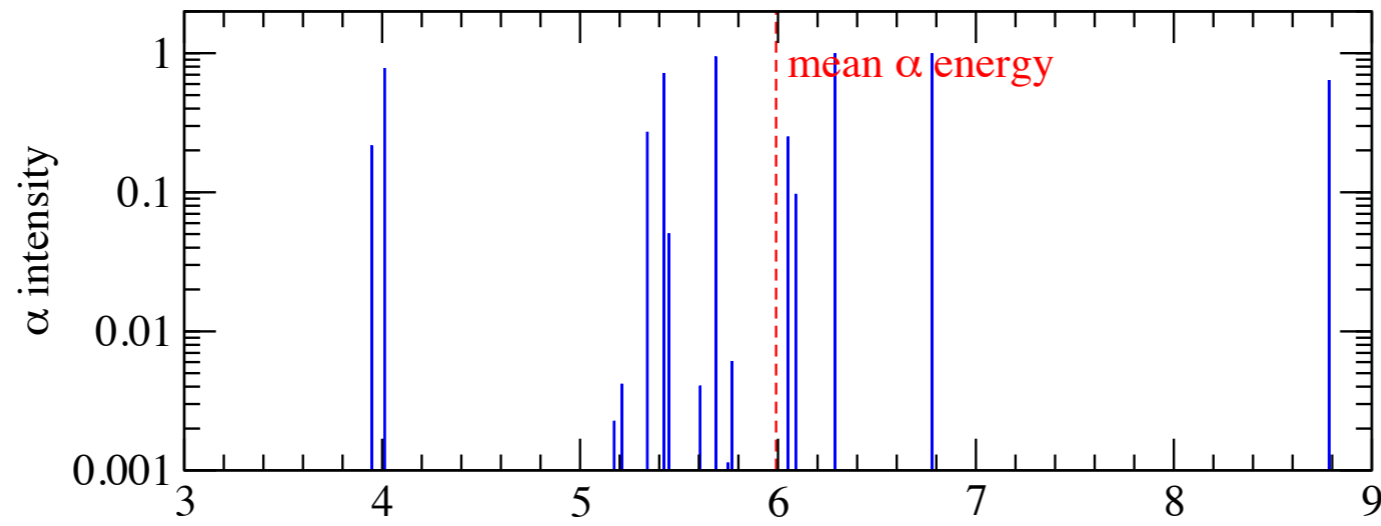
^{232}Th α decay scheme



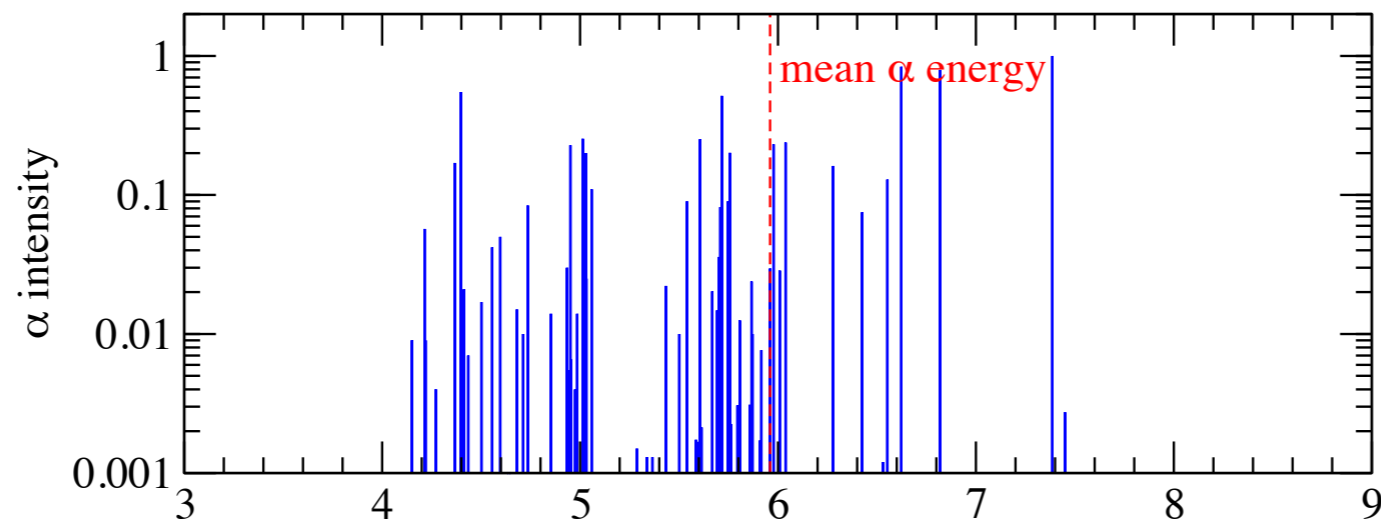
α energies
and
intensities

Natural alpha emission energy spectra

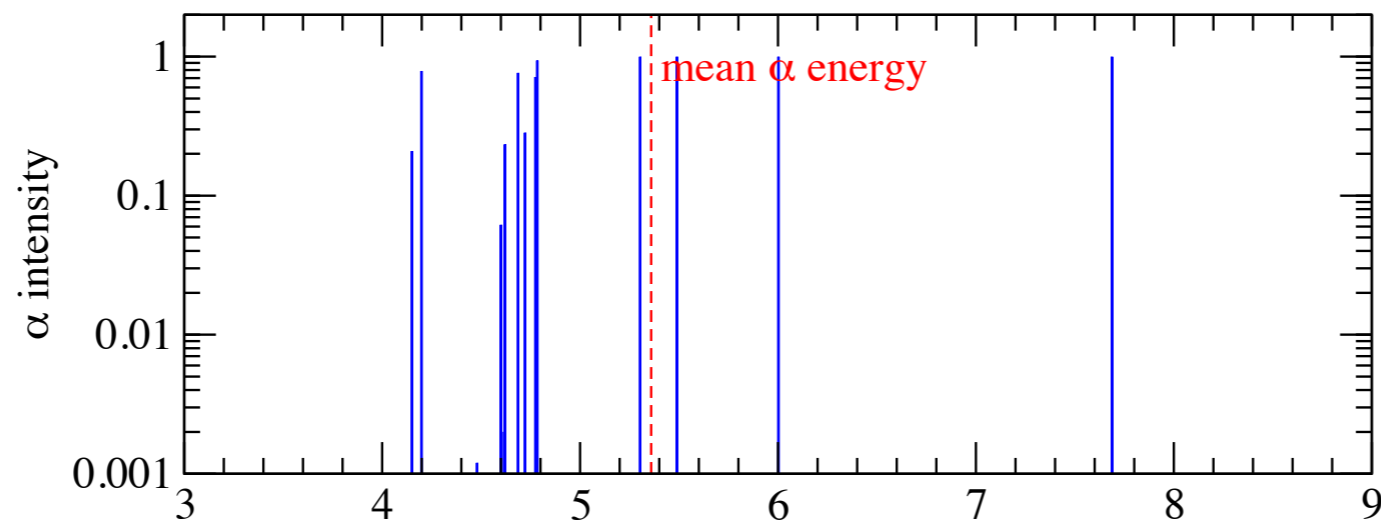
^{232}Th



^{235}U



^{238}U



α 's/decay:

6

7

8

Calculating ^{39}Ar nucleogenic production rate

- α emission from natural radionuclides
- neutron production by α -induced reactions
- $^{39}\text{K}(n,p)^{39}\text{Ar}$

(α, n) neutron production and spectrum

- α emitted at initial energy $E_{\alpha 0}$
- α slows down and eventually stops ($E=0$) in the rock
- before it stops, it can participate in (α, n) reaction

neutron production function

neutrons per 1 α particle

$$P_i(E_{\alpha 0}) = N_i \int_0^{E_{\alpha 0}} \frac{\sigma_{\alpha, n}^i(E_{\alpha})}{\left(-\frac{dE_{\alpha}}{dx}\right)} dE_{\alpha}$$

atomic density
cross section
stopping power

neutron spectrum

$$\frac{dP_i}{dE_n}(E_{\alpha 0}, E_n) = N_i \int_0^{E_{\alpha 0}} \frac{\frac{d\sigma_{\alpha, n}^i(E_{\alpha}, E_n)}{dE_n}}{\left(-\frac{dE_{\alpha}}{dx}\right)} dE_{\alpha}$$

spectrum (or differential c.s.)

(α, n) neutron yield and production rate

neutron production function

neutrons per 1 α particle

$$P_i(E_{\alpha 0}) = N_i \int_0^{E_{\alpha 0}} \frac{\sigma_{\alpha, n}^i(E_{\alpha})}{\left(-\frac{dE_{\alpha}}{dx}\right)} dE_{\alpha}$$

cross section

atomic density

stopping power

neutron spectrum

$$\frac{dP_i}{dE_n}(E_{\alpha 0}, E_n) = N_i \int_0^{E_{\alpha 0}} \frac{\frac{d\sigma_{\alpha, n}^i(E_{\alpha}, E_n)}{dE_n}}{\left(-\frac{dE_{\alpha}}{dx}\right)} dE_{\alpha}$$

spectrum (or differential c.s.)

neutron yield

neutrons per decay of 1 atom of parent nuclide

$$Y_{(\alpha, n)} = \sum_{k=1}^K b_k \sum_{l=1}^{L_k} f_l^{\alpha} \sum_{i=1}^I P_i(E_l)$$

E levels

α decays

targets

branching ratio

level intensity

neutron production rate

neutrons per unit time per unit mass of rock

$$S_n = \lambda \frac{AX N_A}{M} Y$$

elem. abundance

nat. isot. rat.

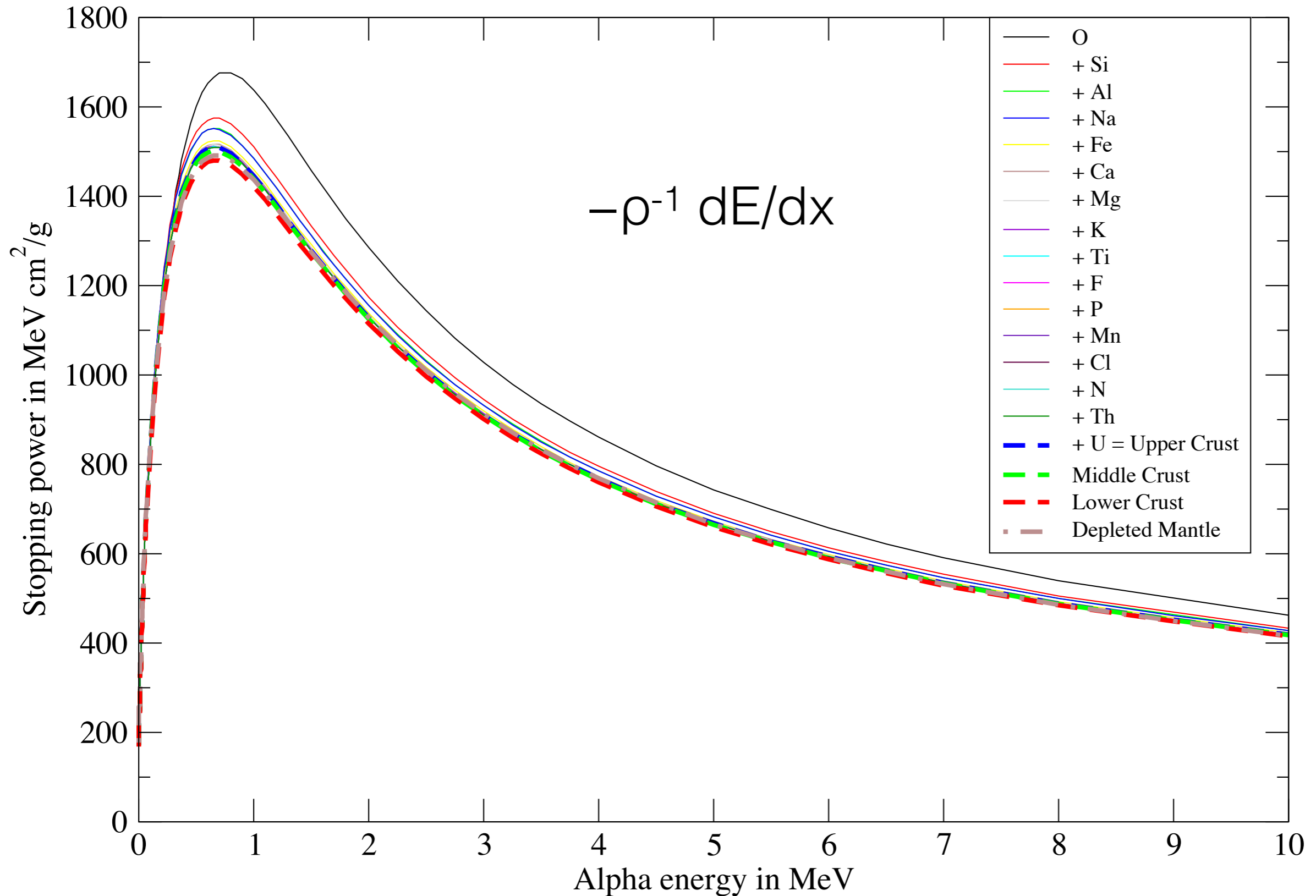
decay constant

atomic mass

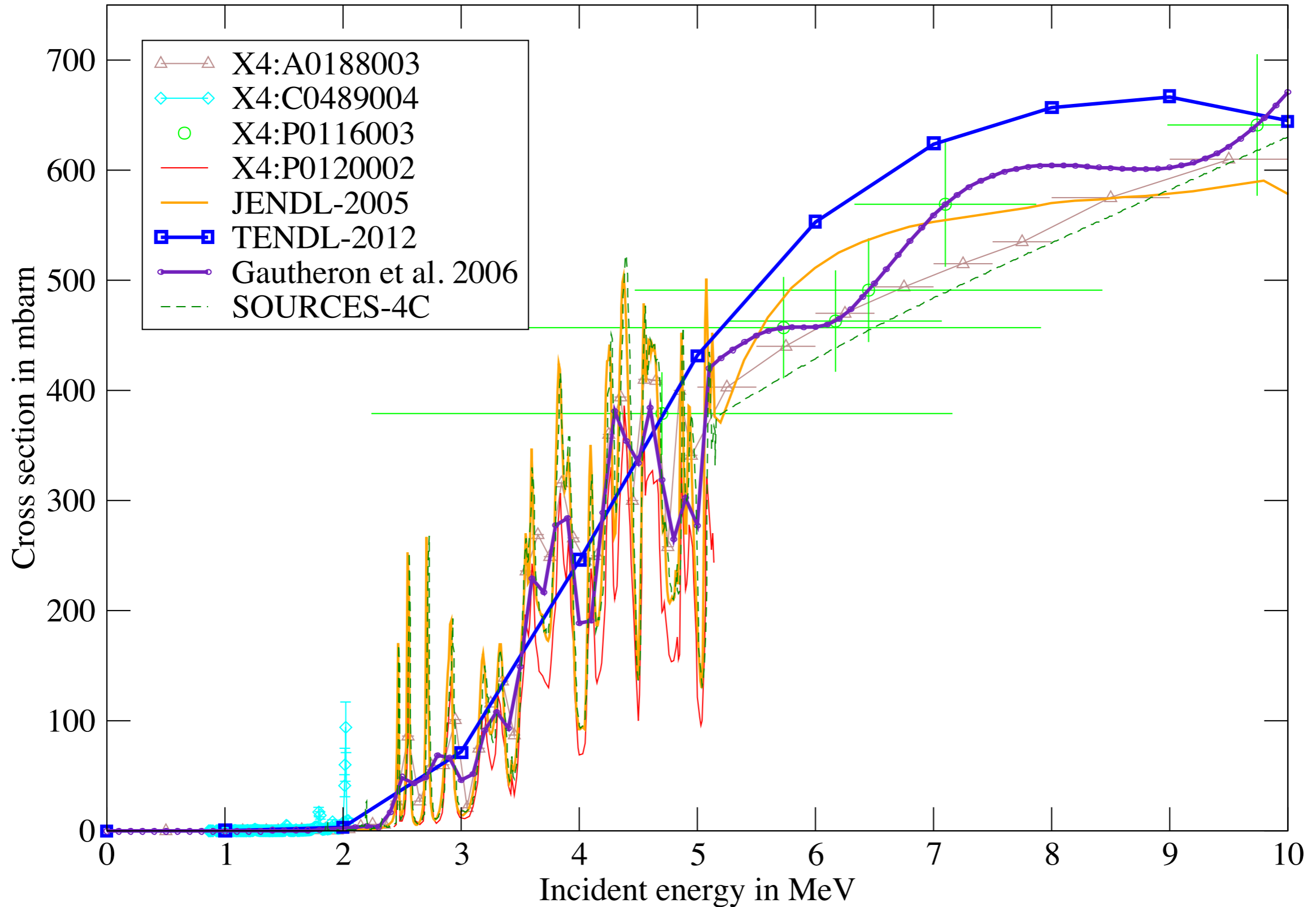
inputs...

Stopping power of alphas in rock

SRIM calc, RG03 / SS04 composition and CRUST2.0 / PREM density (2.70 - 2.88 - 3.05 / 3.42 g/cm³)

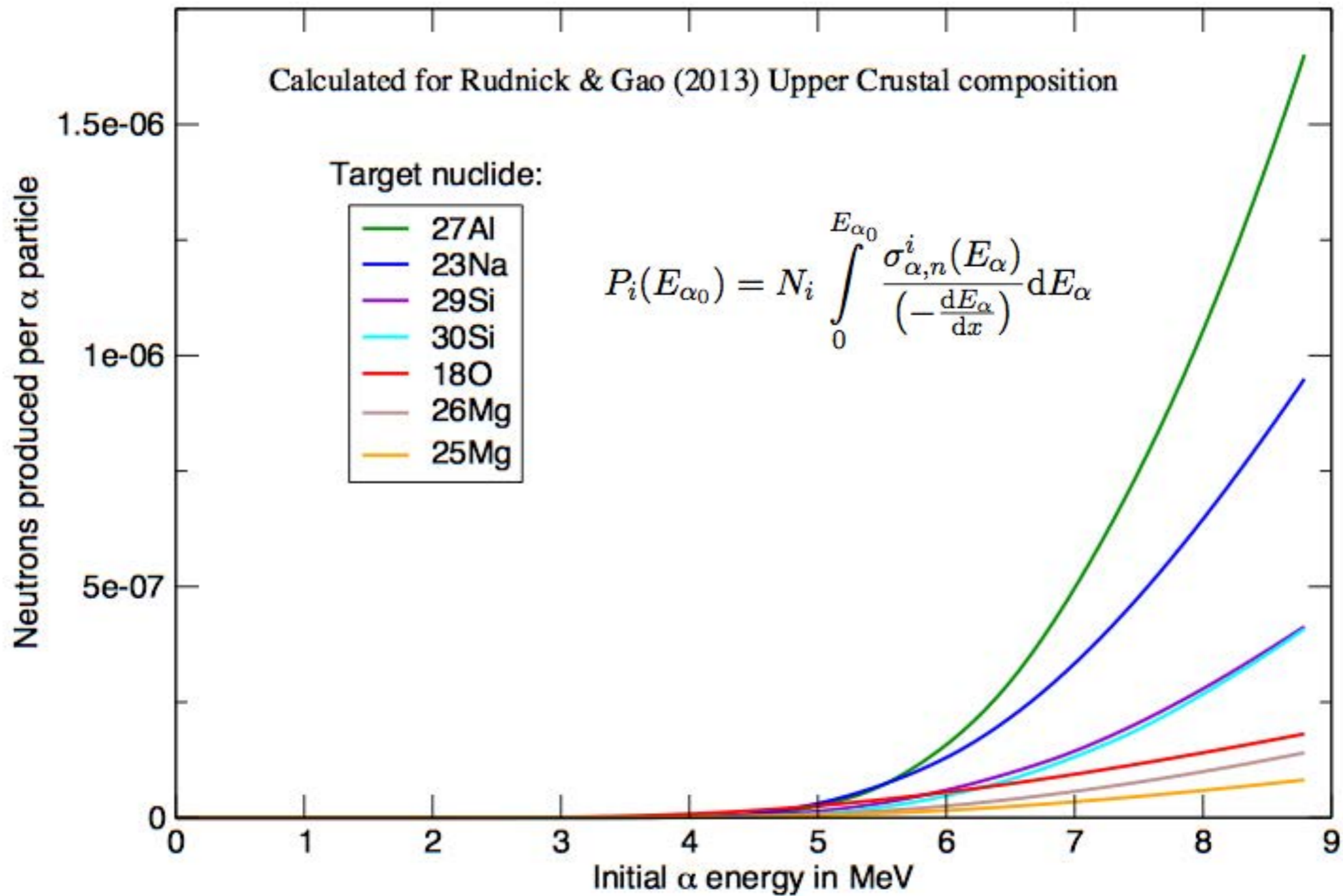


$^{18}\text{O}(\text{a},\text{n})$ cross section



We use: cross sections calculated by TALYS code, <http://www.talys.eu/>

Neutron production function



Calculating neutron spectra

1. Non-relativistic reaction kinematics Ground state-to-ground state

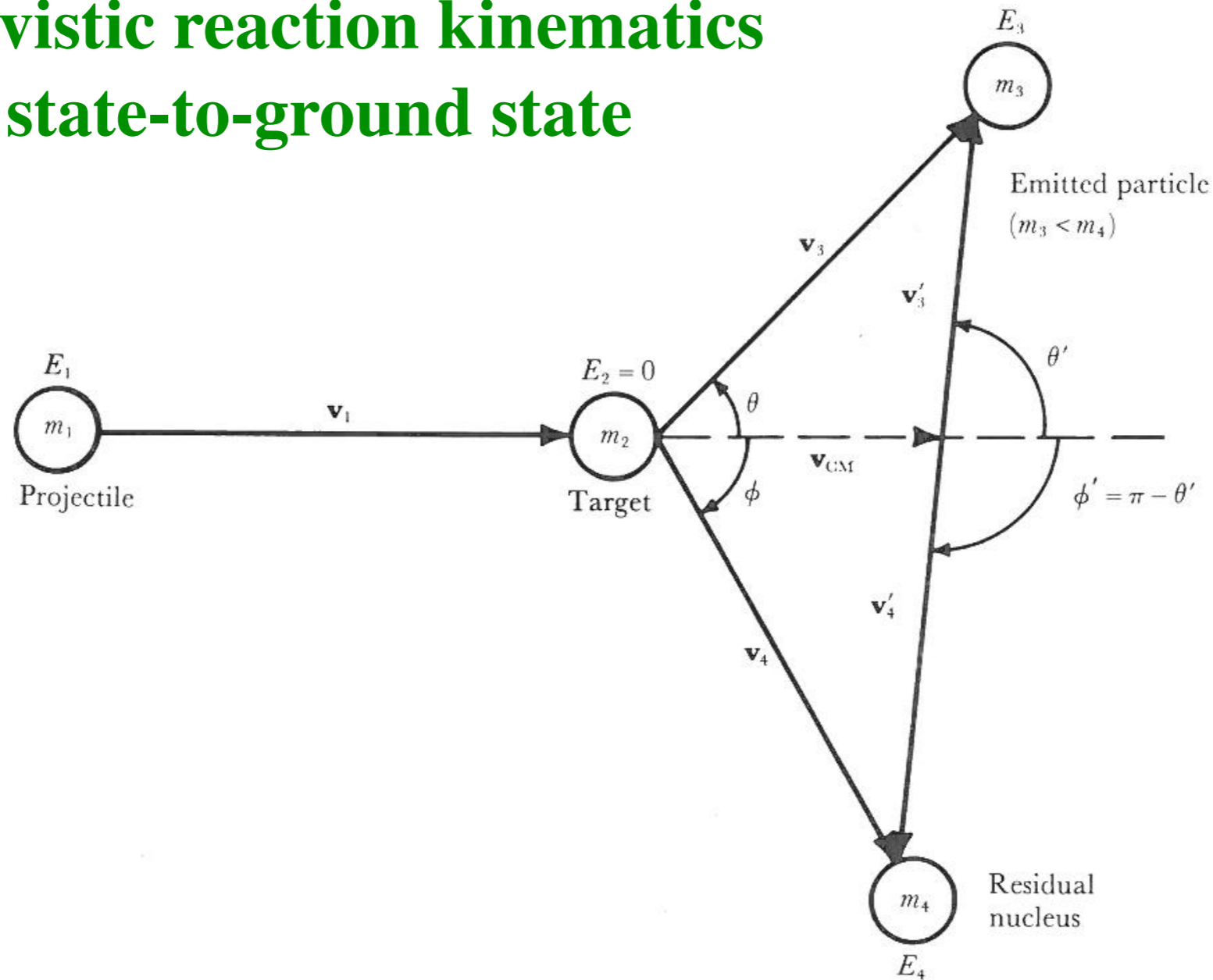


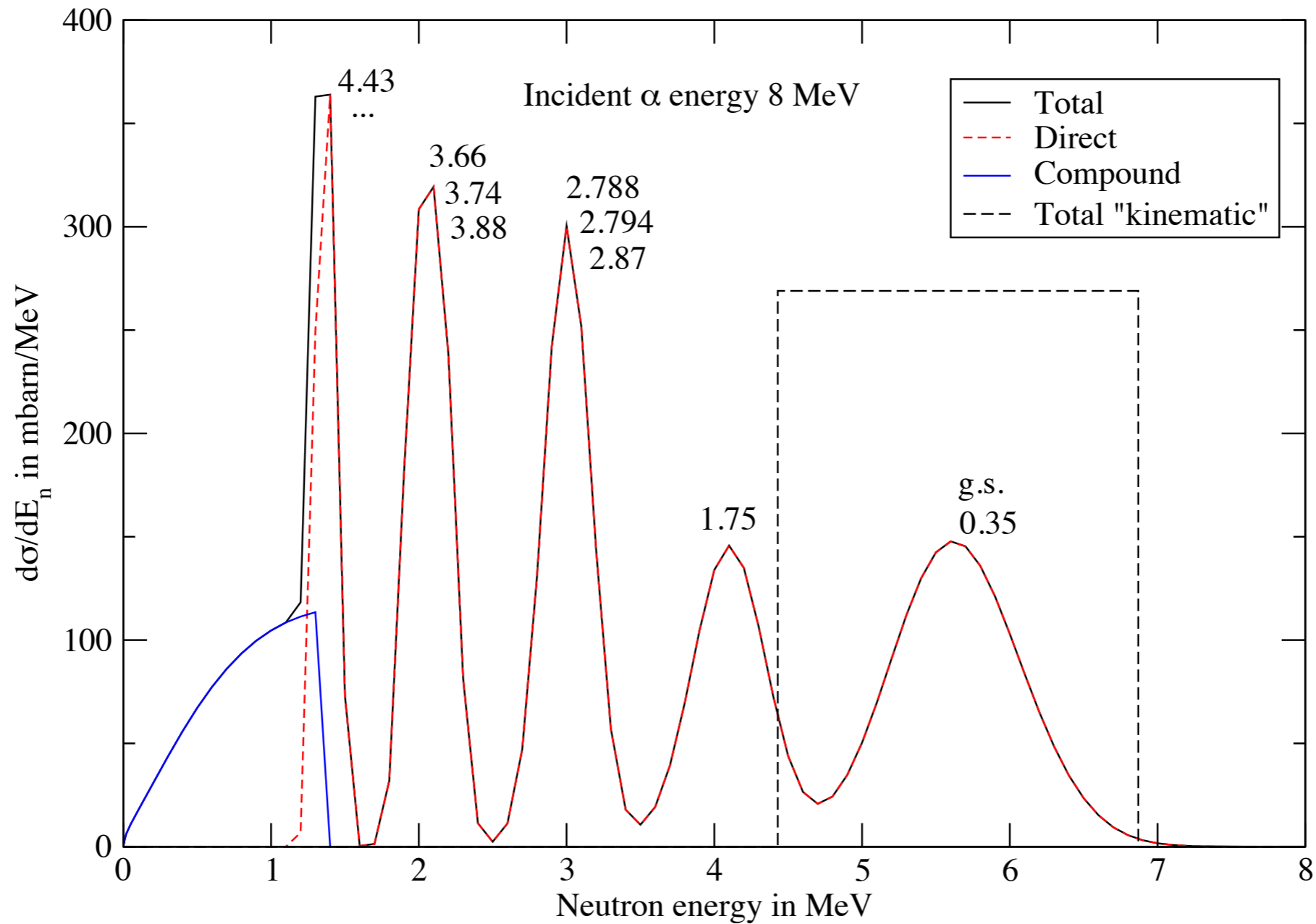
Fig. C-4. Vector diagram showing the relationship of kinematic quantities in the reaction process $m_1 + m_2 \rightarrow m_3 + m_4$.

2. More complete physical picture: TALYS code

Neutron spectrum for 8 MeV $\alpha + {}^{18}\text{O}$

$\alpha + {}^{18}\text{O}$: neutron spectrum

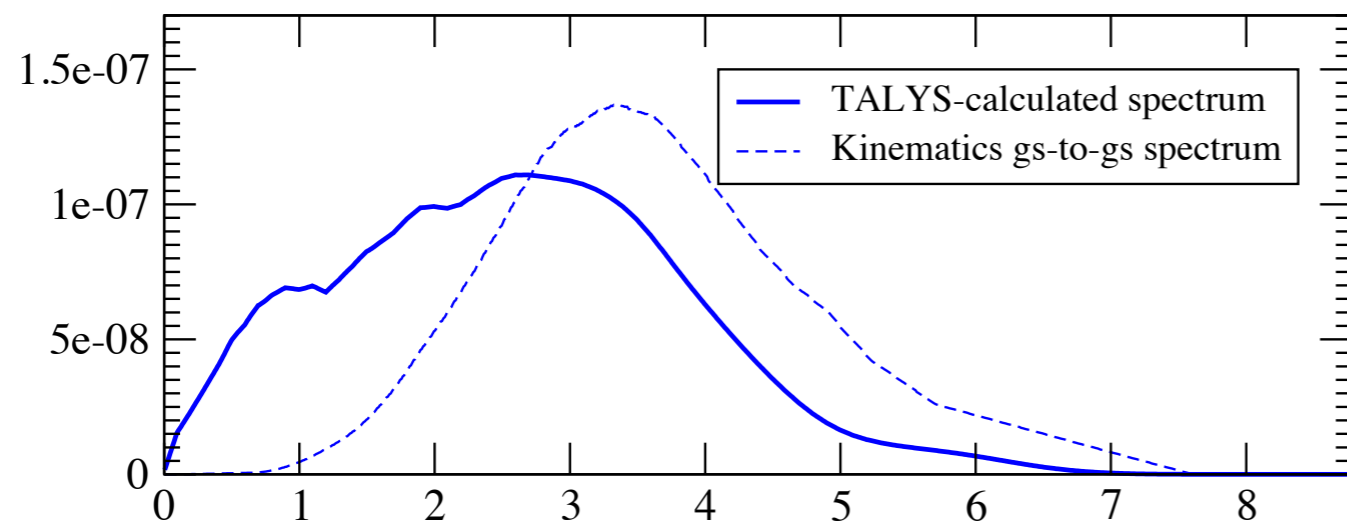
TALYS-1.4



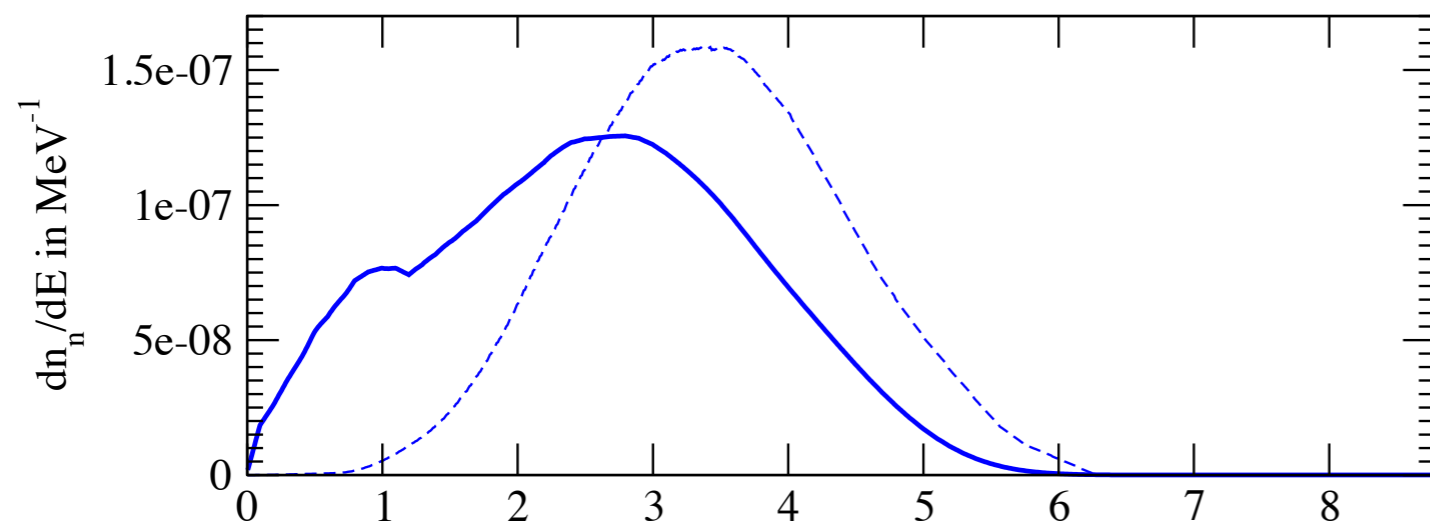
We use: spectra calculated by TALYS code, <http://www.talys.eu/>

$\alpha + {}^{18}\text{O}$: neutron energy spectra

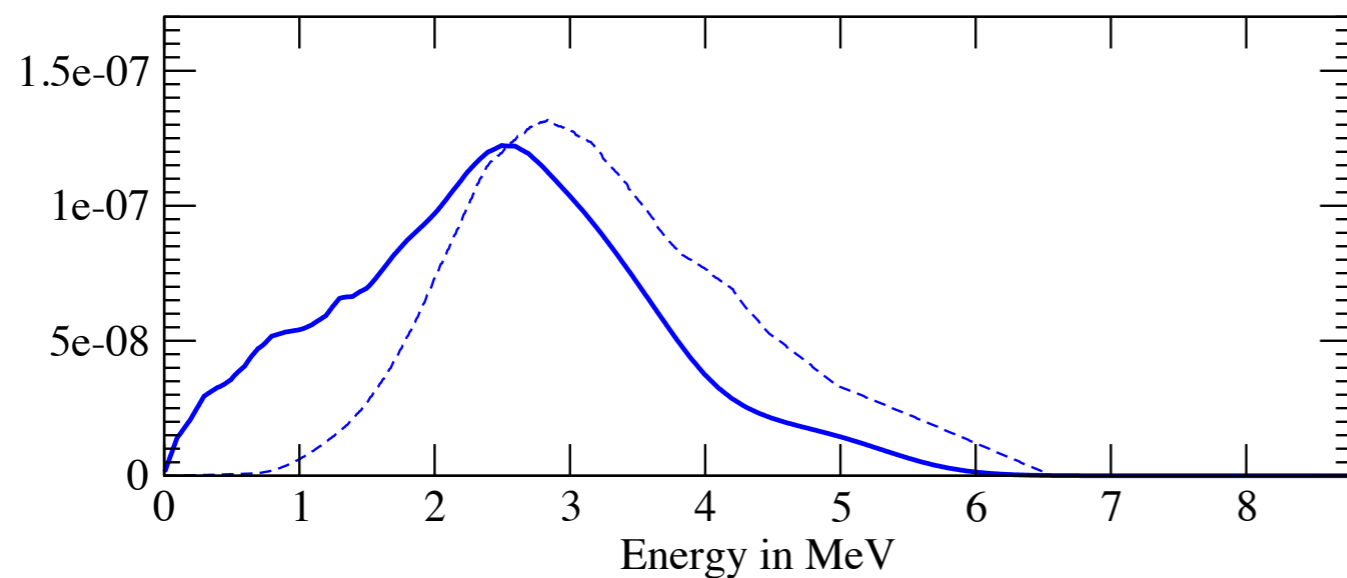
${}^{232}\text{Th}$



${}^{235}\text{U}$



${}^{238}\text{U}$

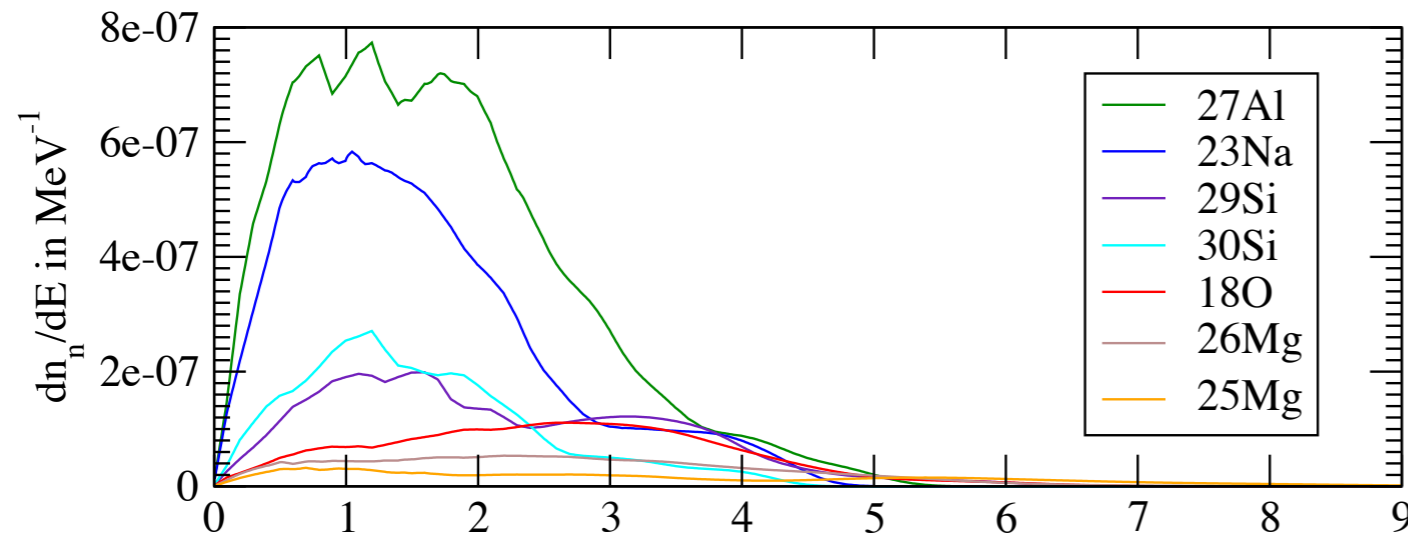


$$\frac{dP_i}{dE_n}(E_{\alpha_0}, E_n) = N_i \int_0^{E_{\alpha_0}} \frac{\frac{d\sigma_{\alpha,n}^i(E_\alpha, E_n)}{dE_n}}{\left(-\frac{dE_\alpha}{dx}\right)} dE_\alpha$$

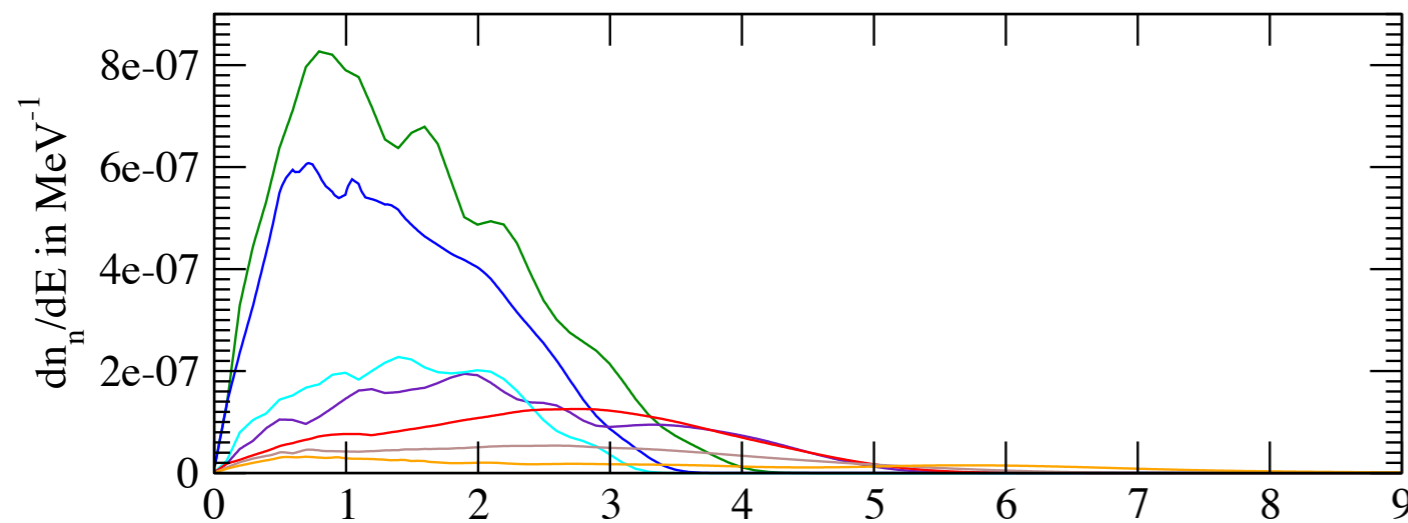
+ sum up over all α -decays,
all α levels

α + target : neutron energy spectra [TALYS]

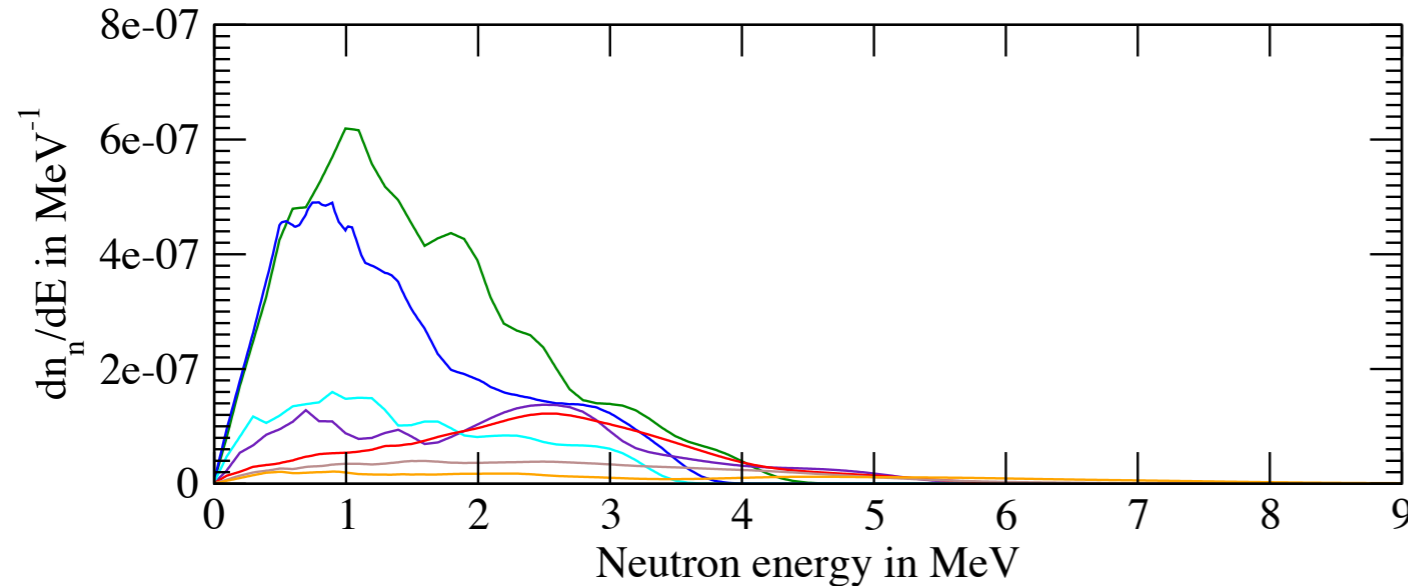
^{232}Th



^{235}U



^{238}U



$$\frac{dP_i}{dE_n}(E_{\alpha_0}, E_n) = N_i \int_0^{E_{\alpha_0}} \frac{\frac{d\sigma_{\alpha,n}^i(E_\alpha, E_n)}{dE_n}}{\left(-\frac{dE_\alpha}{dx}\right)} dE_\alpha$$

+ sum up over all α -decays,
all α levels

Calculating ^{39}Ar nucleogenic production rate

- α emission from natural radionuclides
- neutron production by α -induced reactions
- $^{39}\text{K}(n,p)^{39}\text{Ar}$, also $^{24}\text{Mg}(n,\alpha)^{21}\text{Ne}$

Neutrons propagating and interacting in the rock

- Competition between neutron scattering, various neutron-induced reactions on various targets
- We are interested in specific reactions
- Back-of-the-envelope calculations give us an order of magnitude answer:

$$P_{39}(E_0) = L_n(E_0)\overline{\sigma}_{39}N_{39} \quad (22)$$

where $L_n(E_0)$ is the distance traveled by the neutron until its energy is decreased below the $^{39}\text{K}(n, p)$ threshold, $\overline{\sigma}_{39}$ is the average value of $^{39}\text{K}(n, p)$ cross section, and N_{39} is the atom density of ^{39}K nuclide. Using approximate values,

$$P_{39}(5 \text{ MeV}) \approx 40 \text{ cm} \times 0.3 \text{ barn} \times 9 \times 10^{20} \text{ atoms cm}^{-3} = 0.01 \quad (23)$$

- We use MCNP6 code for a more educated calculation (Monte Carlo N-Particle, <http://mcnp.lanl.gov>)



Results

Table 5: Production rates of ^4He , neutrons, ^{21}Ne , ^{39}Ar .

Composition	Production rates in atoms/kg-yr			
	^4He	neutrons	^{21}Ne	^{39}Ar
<i>RG03</i> , Upper Crust	1.64×10^{10}	10350	0.151	29.9
<i>RG03</i> , Middle Crust	8.98×10^9	6232	0.159	14.7
<i>RG03</i> , Lower Crust	1.53×10^9	1156	0.103	0.792
<i>SS04</i> , Depleted Mantle	2.51×10^7	22.2	0.0380	2.58E-04

Table 6: Production rates of ^{21}Ne by (α, n) and (n, α) and $^{21}\text{Ne}/^4\text{He}$ ratio.

Composition	^{21}Ne prod. rate in atoms/kg-yr			% contrib.		$^{21}\text{Ne}/^4\text{He}$
	(α, n)	(n, α)	Total	(α, n)	(n, α)	
<i>RG03</i> , Upper Crust	863.8	0.15	864	99.98	0.02	5.26×10^{-8}
<i>RG03</i> , Middle Crust	474.7	0.16	475	99.97	0.03	5.29×10^{-8}
<i>RG03</i> , Lower Crust	79.3	0.10	79.4	99.9	0.1	5.18×10^{-8}
<i>SS04</i> , Depleted Mantle	1.18	0.038	1.22	96.9	3.1	4.85×10^{-8}

Yokochi et al. (1997): $^{21}\text{Ne}/^4\text{He} = 4.5 \times 10^{-8}$

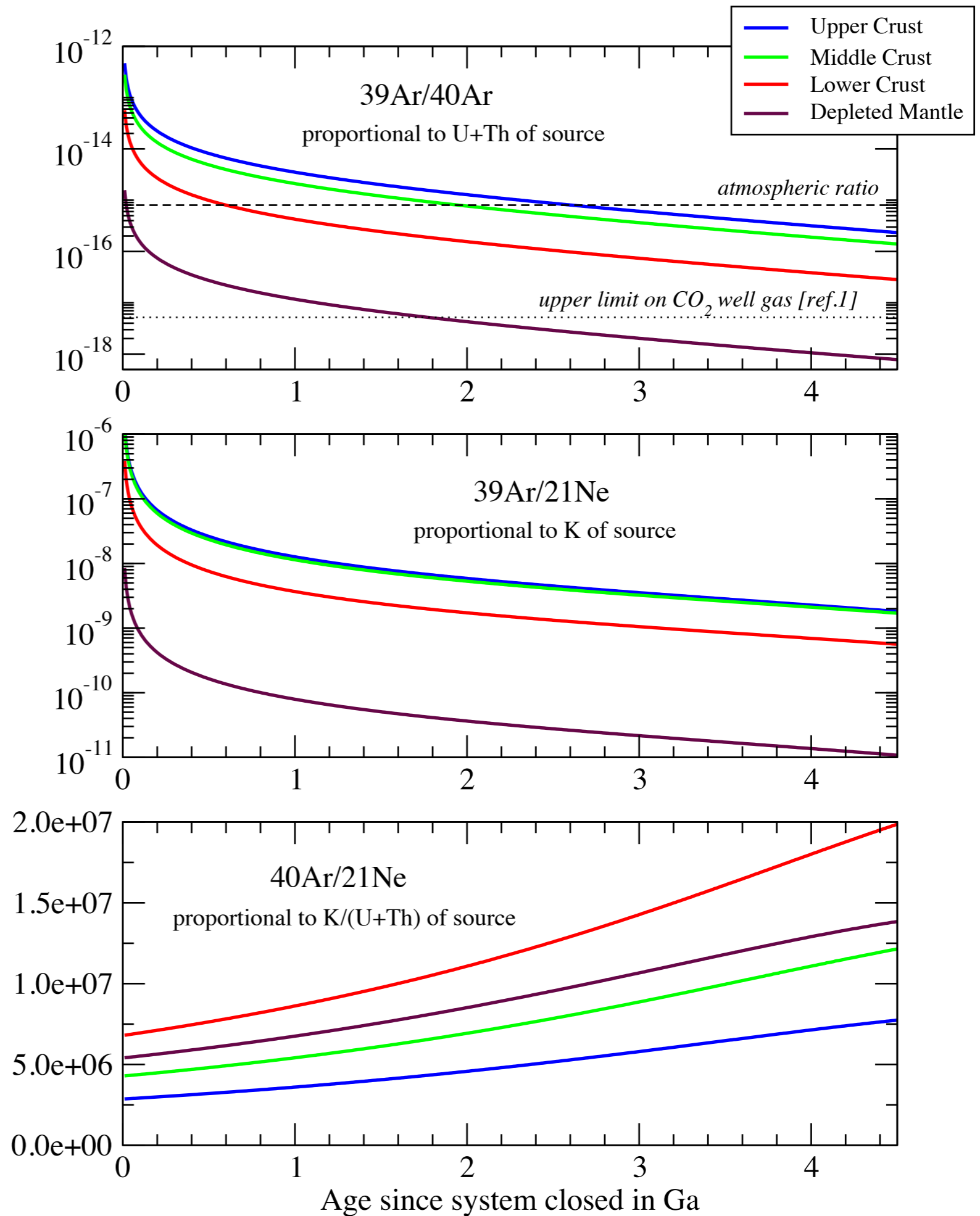
Isotopic ratios

Trade-off between composition and “age” of system

Closed system:

^{21}Ne , ^{40}Ar accumulate

^{39}Ar steady-state level



Summary

- New calculations of nucleonic production rates
- Combination of state-of-the-art nuclear physics tools (TALYS, MCNP6) and my own code to put everything together
- (Re-)Learned aspects of nuclear physics
- Results of interest to hydrology, dark matter experiments, crustal & mantle geochemistry
- We expect the ^{39}Ar detection methods to improve